

A variation of the Pictet–Spengler reaction via a sequential reduction–cyclization reaction of *N*-acylcarbamates: synthesis of 1-substituted tetrahydroisoquinoline derivatives

Chutima Kuhakarn,^{b,*} Nattakan Panyachariwat^b and Somsak Ruchirawat^{a,*}

^aChulabhorn Research Institute, Vipavadee Rangsit Highway, Bangkok 10210, Thailand

^bDepartment of Chemistry, Faculty of Science, Mahidol University, Bangkok 10400, Thailand

Received 18 July 2007; revised 6 September 2007; accepted 13 September 2007

Available online 18 September 2007

Abstract—A new variation of the Pictet–Spengler reaction for the synthesis of 1-substituted tetrahydroisoquinoline derivatives has been developed. The reaction employs the reduction of *N*-acylcarbamates by DIBAL-H followed by simultaneous cyclization mediated by $\text{BF}_3 \cdot \text{OEt}_2$. The synthetic potential of this method has been illustrated by the synthesis of the tetrahydroisoquinoline alkaloids, (\pm)-xylopinine, (\pm)-laudanosine, (\pm)-8-oxo-*O*-methylbharatamine, and (\pm)-isoindoloisoquinolone.
© 2007 Elsevier Ltd. All rights reserved.

Isoquinoline alkaloids are a large family of natural products and display a broad variety of biological activities.^{1a} Among the members of this class of compounds, tetrahydroisoquinoline derivatives constitute a major group. Many of them exhibit important biological activities, for example, anti-inflammatory, anti-microbial, anti-leukemic, and anti-tumor properties.^{1b,c} Accordingly, these have encouraged the development of a number of synthetic methodologies for the construction of the tetrahydroisoquinoline core and have been a subject of several reviews.² The Pictet–Spengler reaction is one of the most effective methods and serves as a convenient method for the construction of racemic as well as asymmetric tetrahydroisoquinoline derivatives and related heterocyclic systems.^{2b,3} The reaction involves the cyclization of imines or iminium ions derived from the condensation of β -arylethylamine derivatives with aldehydes or their synthetic equivalents. The reaction was traditionally carried out in a protic solvent with an acid catalyst.^{4,5} However, the intrinsically poor electrophilicity of the protonated imine means that a variation involving the condensation of *N*-acyl or *N*-sulfonyl β -arylethylamine with aldehydes is often required.⁶ The transformation proceeds by intramolecular electrophilic aromatic substitution by the activated iminium-type intermediate, formed from the aldehyde

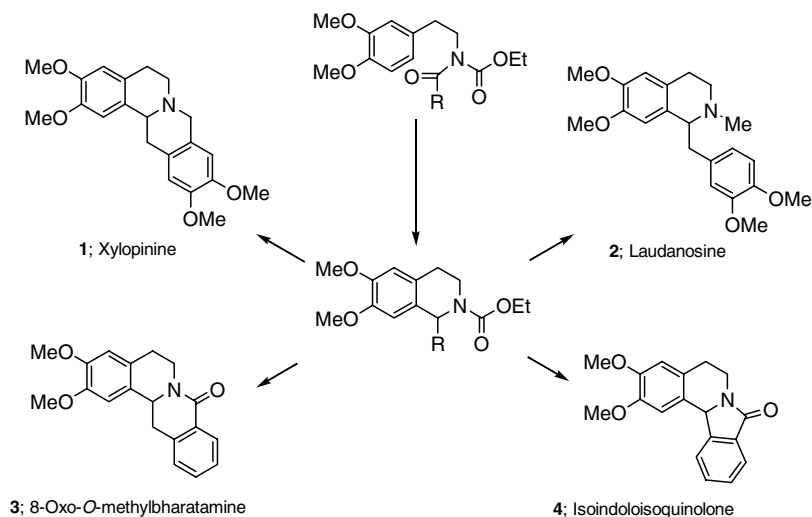
with an activated nitrogen moiety under Lewis acid promotion.

Recently, *N,O*-acetal TMS ethers have been demonstrated to be precursors for the in situ generation of *N*-acyliminium ions.^{7a,b} *N,O*-Acetal TMS ethers can be easily prepared from *N*-acylcarbamates by sequential partial reduction of the amide carbonyl using DIBAL-H followed by TMSOTf trapping of the resulting hemiaminal. In the presence of a suitable Lewis acid, the *N,O*-acetal TMS ethers formed *N*-acyliminium intermediates, which could be trapped with nucleophiles. Synthetic applications of this methodology have also been demonstrated.^{7c,d}

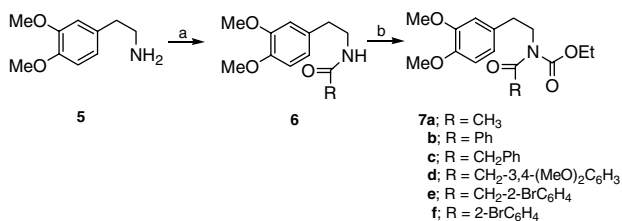
Our research group has long been interested in the development of efficient syntheses of tetrahydroisoquinoline-containing alkaloids.⁸ We herein report an alternative strategy leading to the synthesis of (\pm)-xylopinine (**1**), (\pm)-laudanosine (**2**), (\pm)-8-oxo-*O*-methylbharatamine (**3**), and (\pm)-isoindoloisoquinolone (**4**) by means of a sequential reduction followed by Lewis acid promoted cyclization, in a new modification of the Pictet–Spengler type cyclization process (Scheme 1).

To begin the study, *N*-acylcarbamates **7** were conveniently prepared in two high-yielding steps from 2-(3,4-dimethoxyphenyl)ethylamine (**5**) as shown in Scheme 2. The starting 2-(3,4-dimethoxyphenyl)ethylamine (**5**) was treated with various acid chlorides under

* Corresponding authors. Tel.: +66 2 201 5155; fax: +66 2 354 7151 (C.K.); e-mail addresses: sckkk@mahidol.ac.th; somsak@cri.or.th



Scheme 1. Synthetic plan.

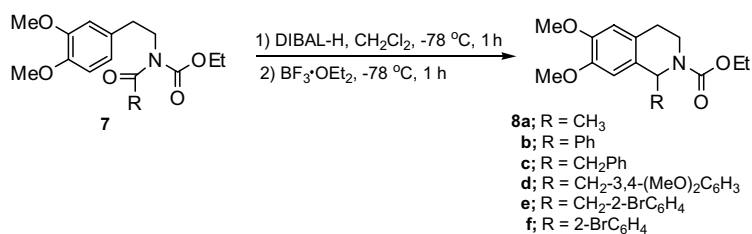
Scheme 2. Synthesis of *N*-acylcarbamate precursors **7**. Reagents and conditions: (a) RCOCl , Na_2CO_3 , 1:1 v/v $\text{CH}_2\text{Cl}_2/\text{H}_2\text{O}$; (b) *n*-BuLi, THF, -78°C then ethyl chloroformate.

biphasic conditions (80–96% yields), and the resulting amides were lithiated with *n*-BuLi followed by trapping with ethyl chloroformate to yield *N*-acylcarbamates **7a–f** with yields ranging from 72% to 98%.

As shown in Table 1, upon treatment of *N*-acylcarbamates **7** with 1.5 equiv of DIBAL-H in hexane followed by sequential addition of $\text{BF}_3\cdot\text{OEt}_2$, smooth production of the 1-substituted tetrahydroisoquinoline derivatives **8** was observed.⁹ Yields were good to excellent and were

generally higher than those previously recorded for acid catalyzed Pictet–Spengler condensation of *N*-acyl or *N*-sulfonyl β -arylethylamines with aldehydes.^{6c,d} Attempts to prepare tetrahydroisoquinoline derivatives from precursors with no activating benzene ring substituents under similar conditions met with failure,¹⁰ the oxygenated aromatic ring is essential for successful cyclization. In contrast to Suh's observations, attempts to trap the hemiaminal formed by the TMSOTf–pyridine system failed to provide the *N,O*-acetal TMS ether, and only fragmentation products were isolated, that is, the corresponding aldehyde and 2-(3,4-dimethoxyphenyl)ethylcarbamic acid ethyl ester. Furthermore, in a pairwise comparison, treatment of 2-(3,4-dimethoxyphenyl)ethylcarbamic acid ethyl ester with phenylacetaldehyde under acid catalyzed Pictet–Spengler conditions (HCO_2H , 110°C , 7 h) gave **8c** in low yield (20%).

Having constructed the tetrahydroisoquinoline core, further elaborations of **8d** to (\pm)-xylopinine (**1**) and (\pm)-laudanosine (**2**); **8e** to (\pm)-8-oxo-*O*-methylbharatamine (**3**) and **8f** to (\pm)-isoindoloisoquinolone (**4**) were straightforward following the previously reported

Table 1. Sequential partial reduction–Lewis acid mediated cyclization of *N*-acylcarbamates **7**

Entry	7; R =	Product (% yield)
1	CH ₃	8a ; 76
2	Ph	8b ; 80
3	CH ₂ Ph	8c ; 98
4	CH ₂ -3,4-(MeO) ₂ C ₆ H ₃	8d ; 83
5	CH ₂ -2-BrC ₆ H ₄	8e ; 83
6	2-BrC ₆ H ₄	8f ; 71

procedures. Treatment of **8d** with triflic anhydride/DMAP efficiently gave (\pm)-8-oxoxylopinine, which underwent reduction by LiAlH_4 to provide (\pm)-xylopinine (**1**) (75%, 2 steps).¹¹ Reduction of **8d** by LiAlH_4 afforded (\pm)-laudanosine (**2**) in 91% yield.^{6d} The alkaloids (\pm)-8-oxo-*O*-methylbharatamine (**3**) and (\pm)-isoidoloisoquinolone (**4**) were obtained in 95% and 78% yields, respectively, by lithium–bromine exchange of **8e** and **8f** using *t*-BuLi, followed by cyclization.^{6d}

In conclusion, we have developed a new variation of the Pictet–Spengler tetrahydroquinoline synthesis in which the *N*-acyliminium intermediates were formed by partial reduction of *N*-acylcarbamates by DIBAL-H followed by sequential addition of $\text{BF}_3 \cdot \text{OEt}_2$. This procedure appears to be useful, in terms of synthetic efficiency, short reaction time, and mild conditions (-78°C). Therefore, the *N*-acylcarbamates serve as convenient precursors for construction of 1-substituted tetrahydroisoquinolines, which can be further elaborated to the synthesis of tetrahydroisoquinoline-containing natural products. Application of this strategy to the diastereoselective synthesis is forthcoming.

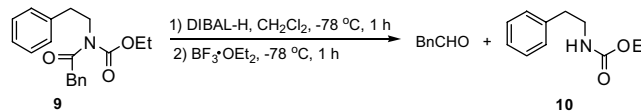
Acknowledgments

We acknowledge financial support from the Thailand Research Fund (TRF) and the Center for Innovation in Chemistry: Postgraduate Education and Research Program in Chemistry (PERCH-CIC). We are also grateful for the facilities provided by the Department of Chemistry, Mahidol University.

References and notes

- (a) Bentley, K. W. *Nat. Prod. Rep.* **2005**, *22*, 249–268; (b) Scott, J. D.; Williams, R. M. *Chem. Rev.* **2002**, *102*, 1669–1730; (c) Iwasa, K.; Moriyasu, M.; Yamori, T.; Turuo, T.; Lee, D.-U.; Wiegrebbe, W. *J. Nat. Prod.* **2001**, *64*, 896–898.
- (a) Bringmann, G.; Ewers, C. L. J.; Walter, R. In *Comprehensive Organic Synthesis*; Trost, B. M., Ed.; Pergamon: Oxford, 1991; Vol. 6, Chapter 4.2, p 733; (b) Chrzanowska, M.; Rozwadowska, M. D. *Chem. Rev.* **2004**, *104*, 3341; (c) Shamma, M.; Moniot, J. L. *Isoquinoline Alkaloids Research, 1972–1977*; Plenum: New York, 1978; (d) Shamma, M. *The Isoquinoline Alkaloids, Chemistry and Pharmacology*; Academic: New York, 1977.
- Cox, E. D.; Cook, J. M. *Chem. Rev.* **1995**, *95*, 1797–1842.
- For recent catalytic Pictet–Spengler reactions: (a) Saito, A.; Takayama, M.; Yamazaki, A.; Numaguchi, J.; Hanzawa, Y. *Tetrahedron* **2007**, *63*, 4039–4047; (b) Manabe, K.; Nobutou, D.; Kobayashi, S. *Bioorg. Med. Chem.* **2005**, *13*,

- 5154–5158; (c) Hegedüs, A.; Hell, Z. *Tetrahedron Lett.* **2004**, *45*, 8553–8555.
- For recent catalytic asymmetric Pictet–Spengler reactions: (a) Seayad, J.; Seayad, A. M.; List, B. *J. Am. Chem. Soc.* **2006**, *128*, 1086–1087; (b) Taylor, M. S.; Jacobsen, E. N. *J. Am. Chem. Soc.* **2004**, *126*, 10558–10559.
- (a) Orazi, O. O.; Corral, R. A.; Giaccio, H. *J. Chem. Soc., Perkin Trans. 1* **1986**, 1977–1982; (b) Silveira, C. C.; Bernardi, C. R.; Braga, A. L.; Kaufman, T. S. *Tetrahedron Lett.* **2003**, *44*, 6137–6140; (c) Lukanov, L. K.; Venkov, A. P.; Mollov, N. M. *Synthesis* **1987**, 204–206; (d) Comins, D. L.; Thakker, P. M.; Baevsky, M. F. *Tetrahedron* **1997**, *53*, 16327–16340; (e) Buchanan, J. G.; Sable, H. Z. In *Selective Organic Transformations*; Thyagarajan, B. S., Ed.; Wiley-Interscience: New York, 1972; Vol. 2, pp 1–95.
- (a) Suh, Y.-G.; Shin, D.-Y.; Jung, J.-K.; Kim, S.-H. *Chem. Commun.* **2002**, 1064–1065; (b) Suh, Y.-G.; Kim, S.-H.; Jung, J.-K.; Shin, D.-Y. *Tetrahedron Lett.* **2002**, *43*, 3165–3167; (c) Shin, D.-Y.; Jung, J.-K.; Seo, S.-Y.; Lee, Y.-S.; Paek, S.-M.; Chung, Y.-K.; Shin, D. M.; Suh, Y.-G. *Org. Lett.* **2003**, *5*, 3635–3638; (d) Jung, J.-W.; Shin, D.-Y.; Seo, S.-Y.; Kim, S.-H.; Paek, S.-M.; Jung, J.-K.; Suh, Y.-G. *Tetrahedron Lett.* **2005**, *46*, 573–575.
- (a) Ruchirawat, S.; Chaisupakitsin, M.; Patranuwatana, N.; Cashaw, J. L.; Davis, V. E. *Synth. Commun.* **1984**, *14*, 1221–1228; (b) Ruchirawat, S.; Tontoolarug, S.; Saha-kitpichan, P. *Heterocycles* **2001**, *55*, 635–640; (c) Ruchirawat, S.; Bhavakul, V.; Chaisupakitsin, M. *Synth. Commun.* **2003**, *33*, 621–625.
- General procedure*: To a solution of *N*-acylcarbamate (1.0 equiv) in dry CH_2Cl_2 (ca. 0.1 M) at -78°C under an argon atmosphere, was added dropwise DIBAL-H (1.0 M solution in hexane, 1.5 equiv). After 1 h, the mixture was treated with $\text{BF}_3 \cdot \text{OEt}_2$ (1.5 equiv) and was stirred at -78°C for 1 h before being quenched with 15% aqueous sodium potassium tartrate (5 mL), and diluted with dichloromethane (5 mL). The mixture was warmed to room temperature and extracted with dichloromethane (2×20 mL). The combined organic layers were washed with water (20 mL), brine (20 mL), dried (Na_2SO_4), filtered, and concentrated (aspirator then vacuo). The crude product was purified by column chromatography (SiO_2 , ethyl acetate/hexanes as eluent) or crystallization.
- Under similar conditions (DIBAL-H, -78°C , 1 h then $\text{BF}_3 \cdot \text{OEt}_2$, -78°C , 1 h), the reaction of compound **9** yielded phenylacetaldehyde (60%) and phenethylcarbamate ethyl ester (**10**) (69%), implying that the reduction took place smoothly but that the cyclization was difficult.



- Davis, F. A.; Mohanty, P. K. *J. Org. Chem.* **2002**, *67*, 1290–1296.